



Potential of Macroalgae for the removal of Aluminum ion and organic pollutants from Natural Waters

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ABSTRACT

Previous studies showed that the immobilized algae gracilaria and sargassum absorbs a percentage of the emergent semi-volatile pollutants found in the Puerto Nuevo River. Epidemiological studies suggest that a relationship exists between Aluminum (Al) and Alzheimer disease (AD) involving relative risks for population exposed to Al concentrations in drinking water higher than 0.1mg/L. This research is based on the 'New Mortality Statistics' of 2008 in Puerto Rico. Three towns with a higher incidence of deaths and three towns with the lowest incidence of deaths related to Alzheimer were chosen and the rivers or reservoirs that supply water to the same. To carry out this research the algae gracilaria and sargassum were used; these are not laboratory algae. Then it was proceeded to the identification of the aluminum in water samples and the immobilization of the algae with silica, for later to precede with the absorption of the aluminum, both as the aluminum substance and the river water samples. Finally the detection of aluminum in the samples was realized by a colorimetric method EDTA Complexometric. The immobilized algae, glacialaria and sargassum, absorbed 67 and 64 % of the aluminum present in the model solutions. Glacialaria spp. packing was more effective removing the aluminum present in the solutions. According to our previous and current research results we can conclude that macroalgae packing are a green and cheap alternative to remove simultaneously organic pollutants and Aluminium from water samples.

INTRODUCTION

- Aluminum is the most abundant metal on the planet. It is mostly found combined with other elements. Although we are in contact with it everyday, it can be very toxic when exposed to certain amounts.
- Epidemiological studies suggest that a relationship exists between Aluminum (Al) and Alzheimer disease (AD) involving relative risks for population exposed to Al concentrations in drinking water higher than 0.1mg/L.
- Because marine environments are normally scarce in these metals, some marine algae especially have developed efficient mechanisms to gather these heavy metals from the environment and take them up.
- Algae have different type of uses when it comes to the term of immobilization; one way to immobilize algae is with silica. A great way to use immobilized algae is by absorbing liquid pollutants. (Hameed & Hammouda, 2007). In recent studies we demonstrated that immobilized algae were effective alternative to remove organic pollutants from water.
 - ✓ Can the immobilized algae, *Glacilaria* spp. and *Sargassum* spp. , simultaneously remove the organic pollutants and the aluminum present in water samples?



METHODOLOGY

ALUMINUM

Macroalgae SPE capacity to remove Al

- Water samples were collected and acidified
- 1 L was filtrated
- And pre concentrated to 100 mL

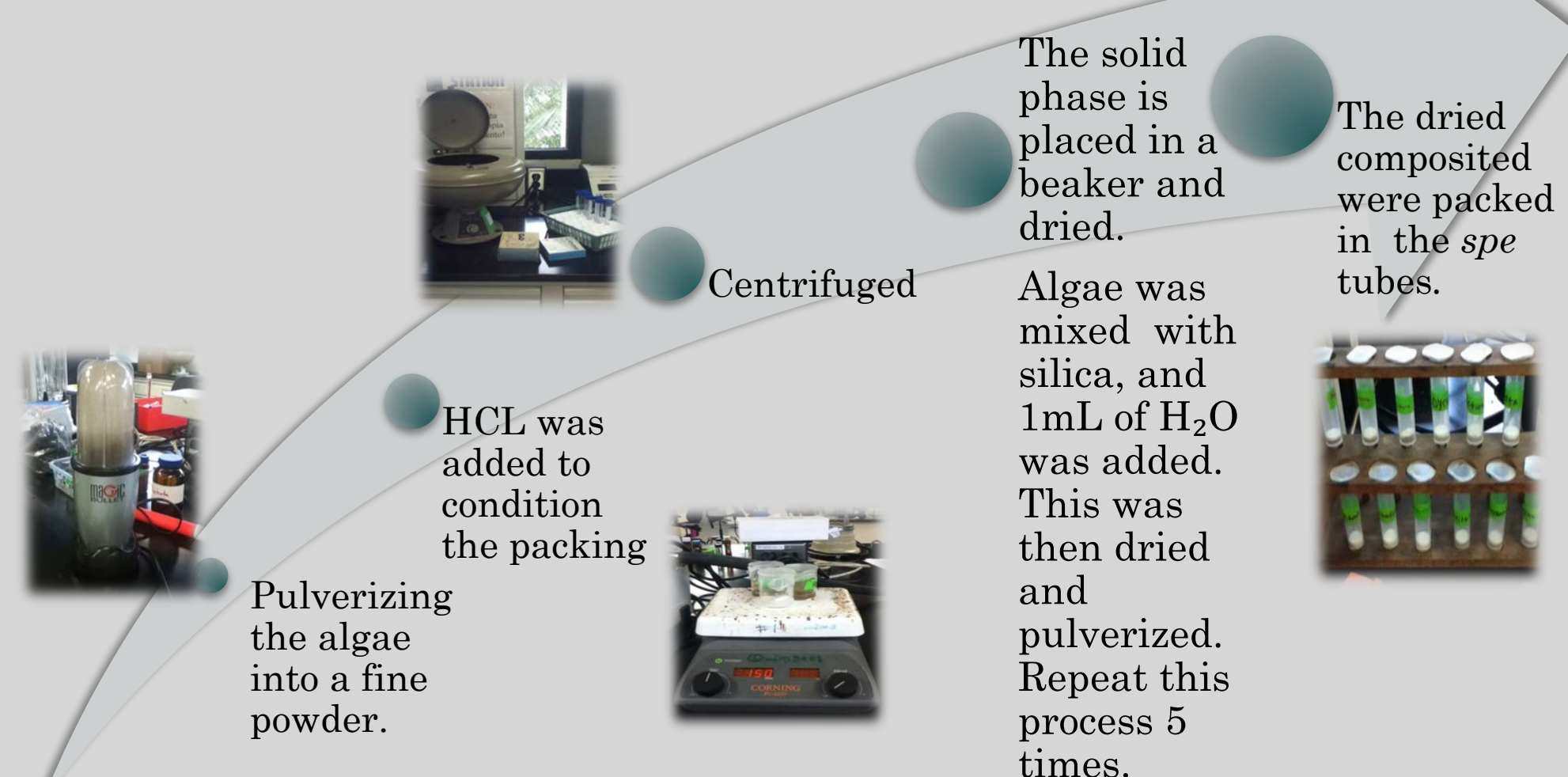
Sample Preparation for Al testing

- HCl was used to Conditioning the algae-SPE tubes
- An Aluminum standard solution of 5ppm was used to determine the algae-spe potential to remove Al from water samples

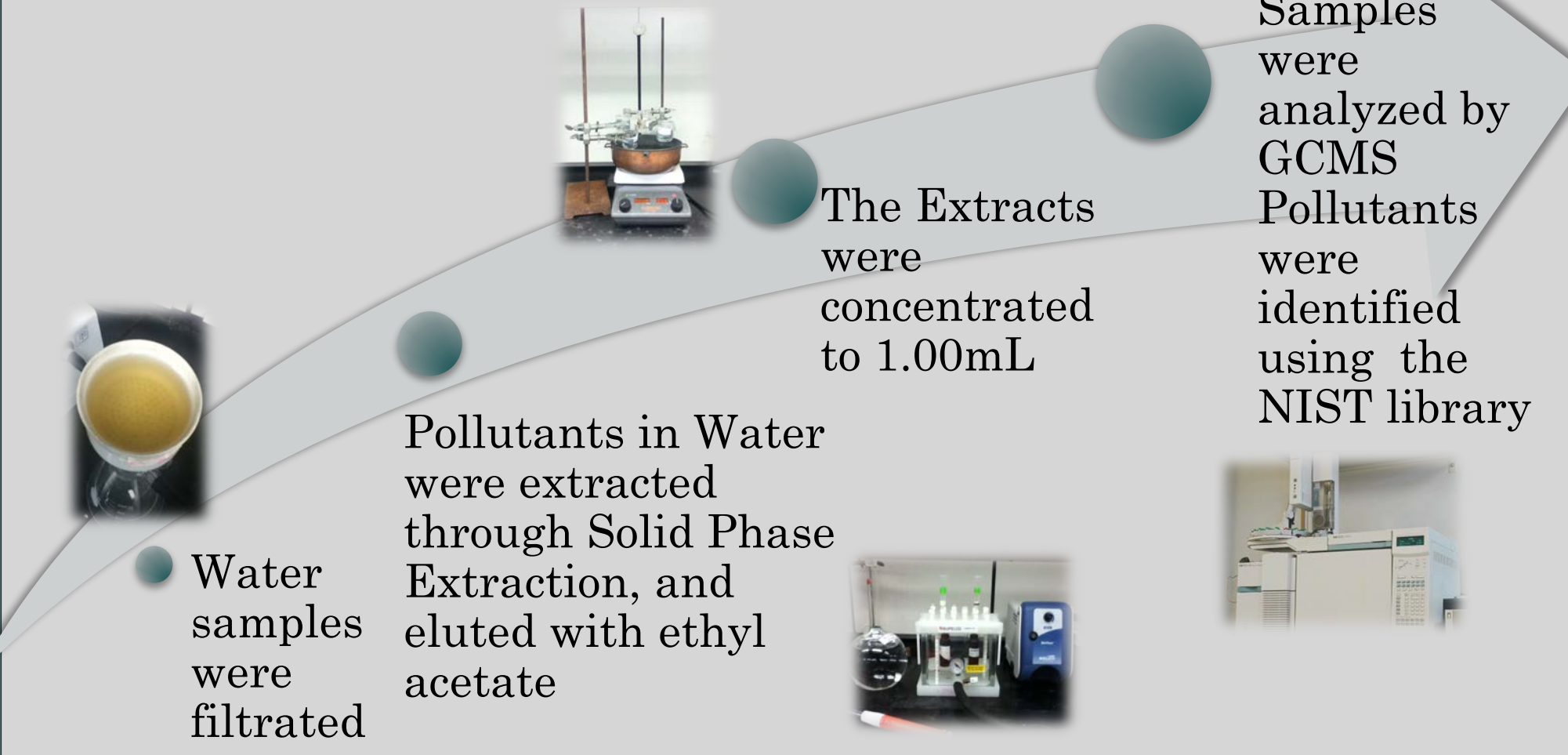
- A calibration curve was made with an aluminum solutions in the range of 0.1 to 5ppm.
- Solutions were made for each sample
- Samples were measured in the Ultraviolet Spectrometer (UV)

Detection of Aluminum by colorimetric method EDTA Complexometric

Preparation of Macro algae Filters

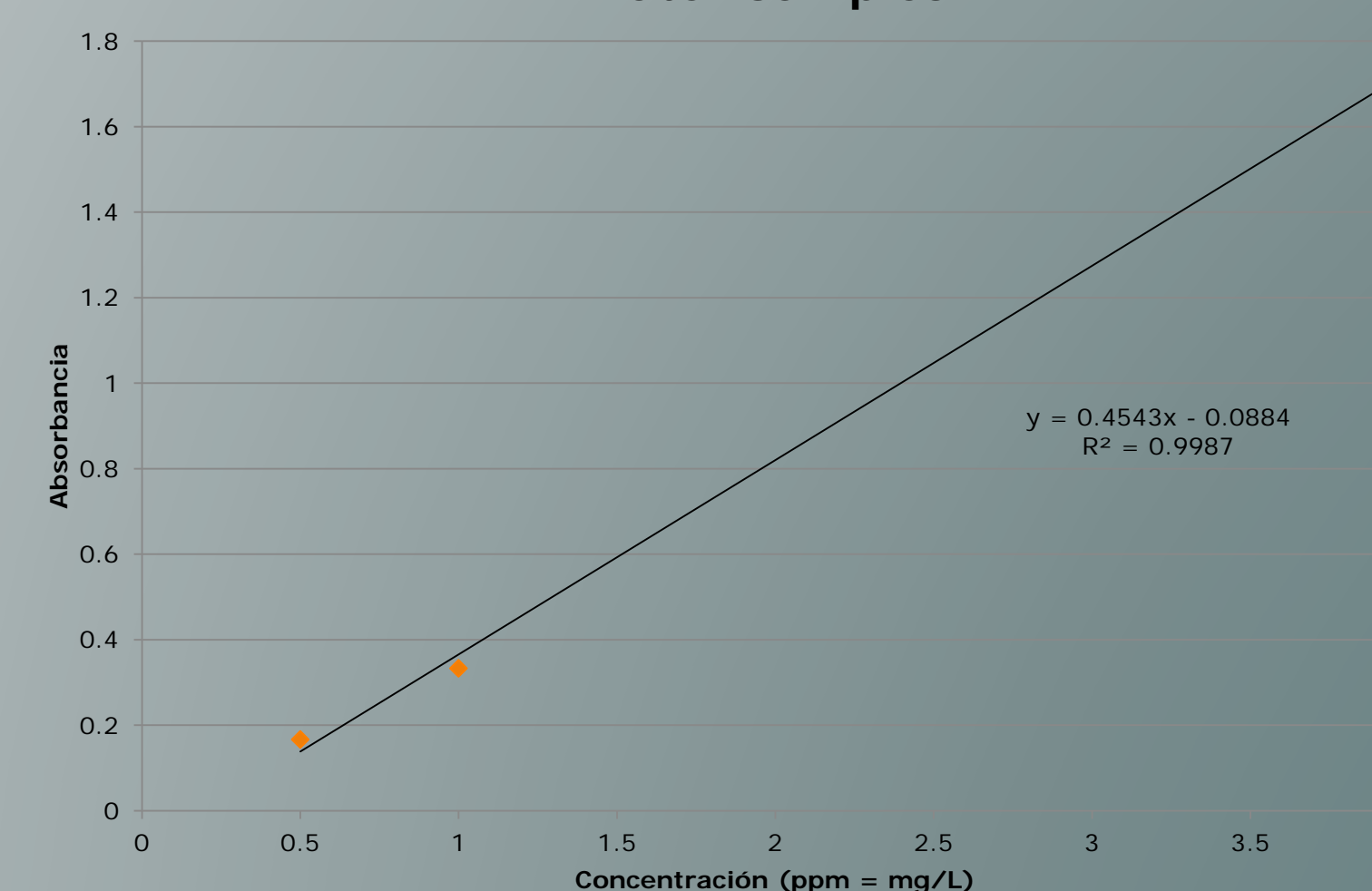


GCMS Analysis of Water Samples – for organic pollutants



RESULTS

Calibration Curve for the Determination of Al in Water Samples



Macro-Algae SPE Al Retention Capacity

Algae	Absorbance (±0.0001)	Al Absorbed (mg/L)	Residual Al Concentration (ppm = mg/L)	Absorption Capacity (µg Al/ g dry algae)
<i>Glacilaria</i>	0.6667	3.34	1.66	33
<i>Sargassum</i>	0.7333	3.20	1.80	32

Aluminum Concentration in Water Samples

Sites	Absorbance (±0.0001)	Concentration (ppm = mg/L)
Río Grande de Loíza	0.2667	0.078
Carraizo	0.3000	0.085
Pitahaya	0.2667	0.078
Maunabo	0.2667	0.078
Blanco	0.4000	0.175

GCMS analysis of Semi volatile Organic Pollutants

Compound	Column 1 Ret Time	Column 2 Area	Compound	ret time	area
Ethylbenzene	5.112	4069201	Ethylbenzene	5.109	3339778
Biciclo	5.776	5.1E+07	Biciclo	5.776	5.1E+07
Dodecanoic acid, methyl ester	21.665	5838195	Dodecanoic acid, methyl ester	22.849	8507281
Phenol, 2,6-bis(1,1-dimethylethyl)	23.5	11289762	Phenol, 2,6-bis(1,1-dimethylethyl)	23.449	7115729
Hexadecane	25.655	10062722	Hexadecane	24.663	4857844
Benzene, 1,1'-(1,3-propanediyl)bis	27.061	6039624	Benzene, 1,1'-(1,3-propanediyl)bis	25.652	3805638
Benzene, 1,1'-(1,3-propanediyl)bis	27.061	6039624	Benzene, 1,1'-(1,3-propanediyl)bis	25.71	6141004
Octadecane	31.908	24789094	Octadecane	26.745	3804888
Hexacosane	36.698	12088449	Hexacosane	27.768	6058721
Tetracosane	40.06	15721882	Tetracosane	30.053	11E+07
Eicosane	41.644	10560399	Eicosane	33.045	6333523
			Eicosane	36.698	4233377

- The organic compounds from the sites are:
 - Ethylbenzene occurs from the use of consumer products, gasoline, pesticides, solvents, carpet glues, varnishes, paints, and tobacco smoke.
 - Phenol is used primarily in the production of nylon and other synthetic fibers.
 - Hexadecane is from diesel.

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CONCLUSIONS

- The immobilized algae, glacialaria and sargassum, absorbed 67 and 64 % of the aluminum present in the model solutions.
- *Glacilaria* spp. packing was more effective removing the aluminum present in the solutions.
- According to our previous and current research results we can conclude that macroalgae packing are a green and cheap alternative to remove simultaneously organic pollutants and Aluminium from water samples.